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products is a mature technology. Most common applications of hydrodeoxygenation are conversion of vacuum gas oil (VGO), atmospheric VGO, shale oil or bitumen to heavier hydrocarbon products and production of middle distillates from these feedstocks. The feedstocks may be dewaxed and hydrotreated prior to hydrodeoxygenation. The hydrodeoxygenation process is normally integrated with hydrogenation and dehydrogenation steps to produce saleable lower boiling hydrocarbon product streams. The hydrodeoxygenation reaction may be performed in the liquid phase or the gas phase, with catalysts comprising zeolite and non-zeolite molecular sieves (NMS) combined with metals such as nickel, cobalt, molybdenum, tungsten, iron or combinations of two or more. Several patents disclose hydrodeoxygenation processes with cobalt as the active metal catalyst. “Dehydrodeoxygenation of heavy hydrocarbon oils: A review” by J. E. Twiggs, Handbook of Petroleum

Refining Processes, Marcel Dekker Inc., 1996, p. 258. The catalysts generally contain nickel and molybdenum, and/or boron. “The use of boron modified zeolites to deoxygenate hydrocarbons” by X. Moulin-Fournier, J. Matravers, C. Seliouine, Eds. Transport and Catalysis, Elsevier, 1994, p. 1203.

Boron-modified zeolites are a well-known class of NMS catalyst. They are prepared by a process that involves modifying a zeolite with NaOH, a borate compound and an alkylating agent such as trimethyl borane. The process development has focused on optimizing the properties of the catalysts used. Once the catalyst is identified, the reaction conditions must be optimized to minimize reactor fouling or coke formation. These reaction conditions include H₂/feed ratio (i.e., molar ratio of H₂/feedstream), temperature (i.e., reaction temperature and/or boiling temperature of the feedstream), contact time, the feed composition itself (i.e., feedstock),

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